

Cross-Disciplinary Science Forum

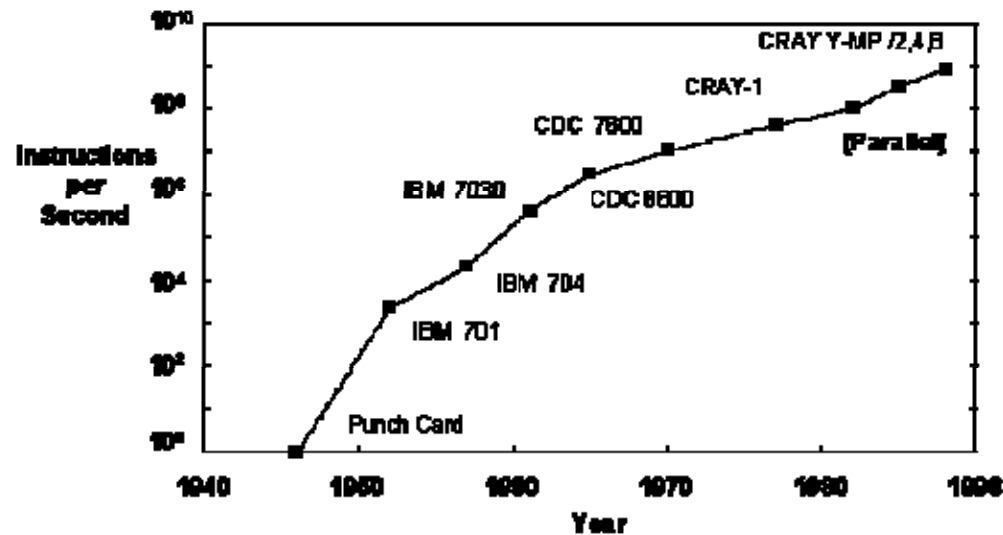
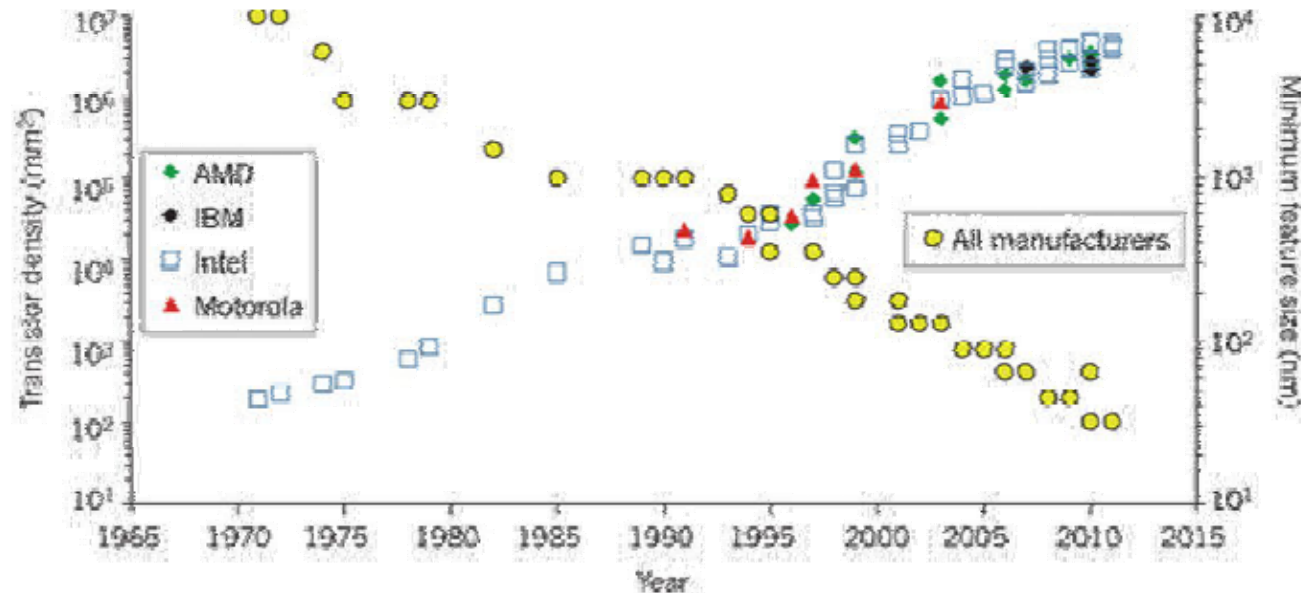
二維的石墨烯半導體

褚志彪

中研院原分所

Moore's Law:

the number of components in integrated circuits had doubled every year from the invention of the integrated circuit in 1958 until 1965 and predicted that the trend would continue "for at least ten years"





The Nobel Prize in Physics 2010

Andre Geim, Konstantin Novoselov



Photo: U. Montan

Andre Geim



Photo: U. Montan

Konstantin Novoselov

The Nobel Prize in Physics 2010 was awarded jointly to Andre Geim and Konstantin Novoselov *"for groundbreaking experiments regarding the two-dimensional material graphene"*

石墨烯的特性

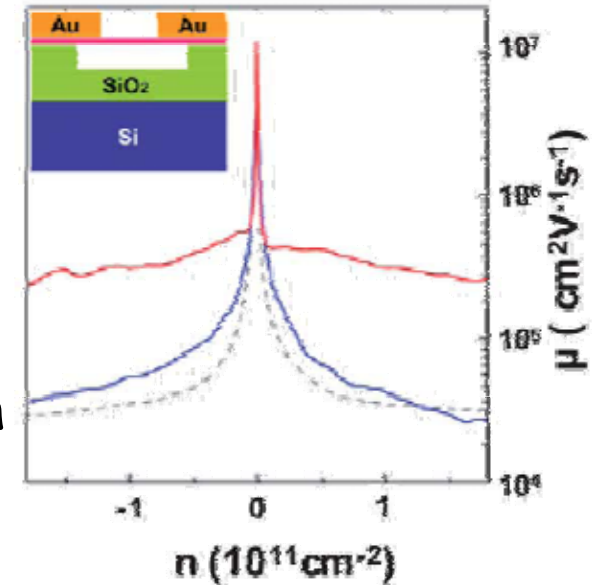
High mobility : > 200000 cm²/Vs

InSb 77000 cm²/Vs
Si 1400 cm²/Vs

$$\frac{1}{\mu_{EFF}} = \frac{1}{\mu_n} + \frac{1}{\mu_{Bal}}$$

$$\mu_n \propto \lambda \quad \text{mean free path}$$

$$\mu_{Bal} \propto L \quad \text{Device length}$$



Bolotin et al *Solid State Communications* **146** 351 (2008)

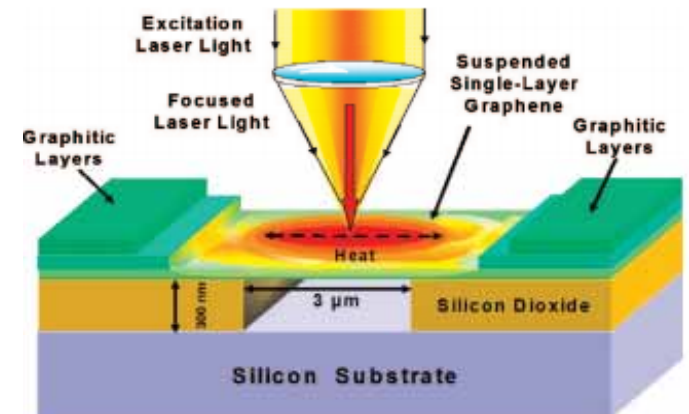
Ultra high mobility is not necessary in scaled graphene device

Superior Thermal Conductivity:

~ 5 × 10³ W/mK

dissipate heat quickly, small thermal vibration Effect

Si: 0.7 ~ 3 Wm⁻¹K⁻¹



Balandin et al *Nano Letters* **8** 902 (2008)

Low Electric Resistivity:

$$\rho = 1.0 \mu\Omega \cdot \text{cm}$$

$$\text{銅: } 1.72 \mu\Omega \cdot \text{cm}$$

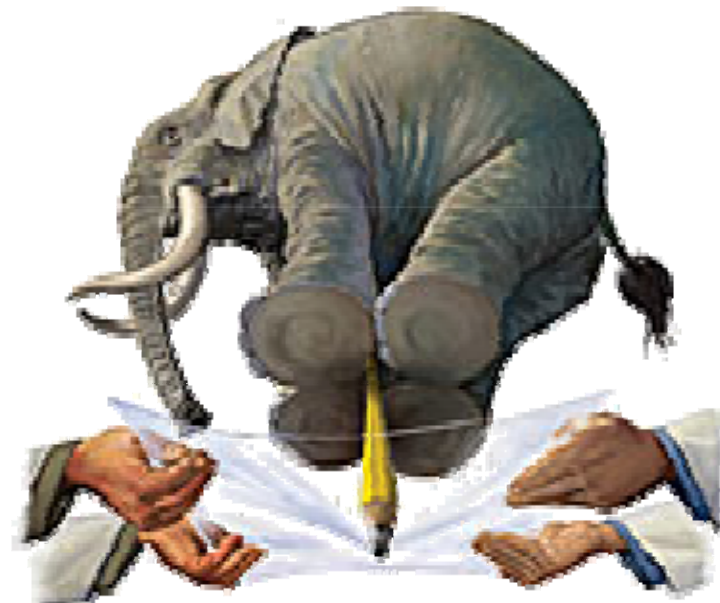
Extreme Tensile strength:

~130 G pascals

$$A \geq \frac{W_{\text{大象}}}{T_{\text{石墨烯}}} = \frac{7000\text{Kg} \times 10 \frac{m}{s^2}}{130 \times 10^9 \frac{N}{m^2}} \approx 0.5 \text{mm}^2$$

Flexibility:

Single-atom-thick sheets, minimum device size.



Ambipolar Electric field effect:

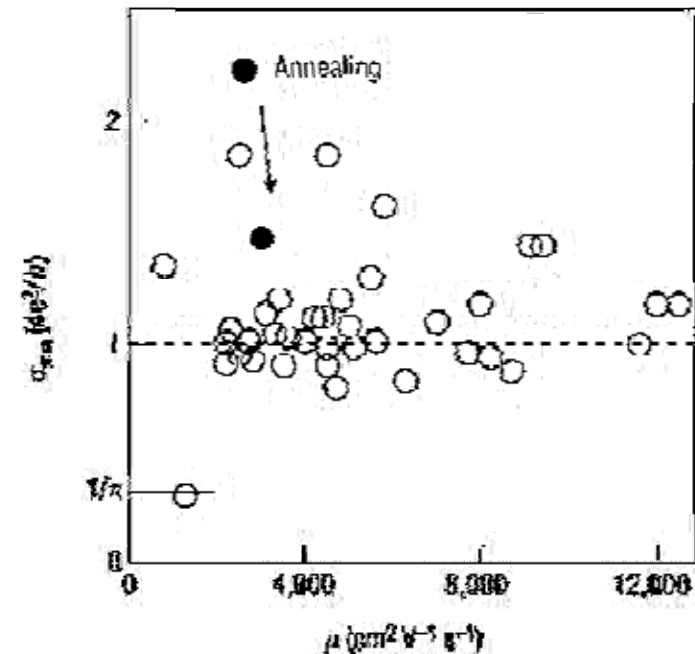
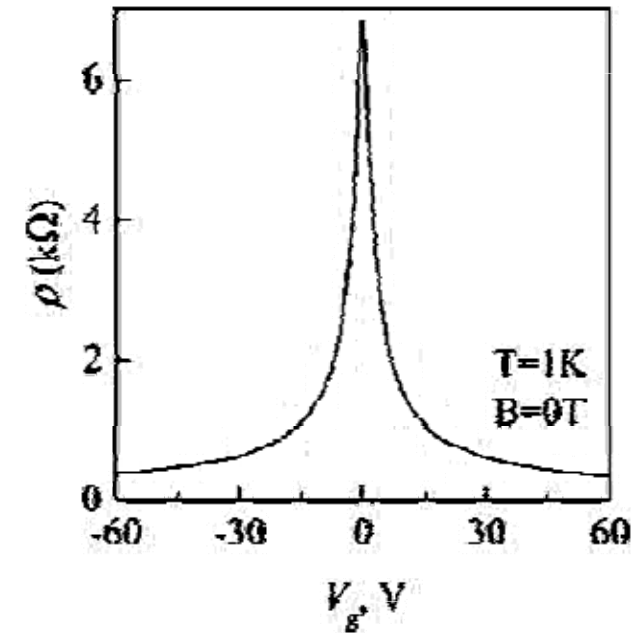
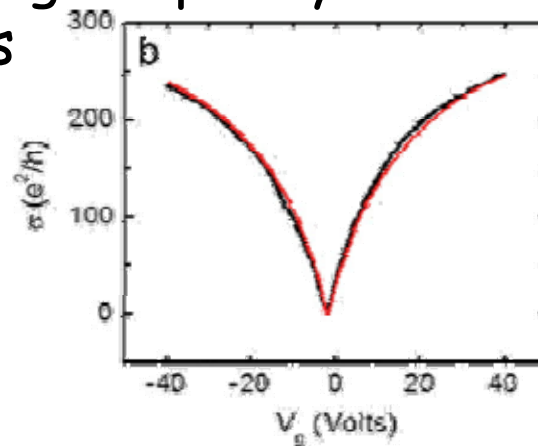
Charge carriers can be tuned continuously between electrons and holes.

Minimum quantized conductivity:

$4e^2/h$ @ E-field = 0

Due to electron-hole puddles
(carrier-density inhomogeneities)

Originated from charge impurity or
surface corrugations

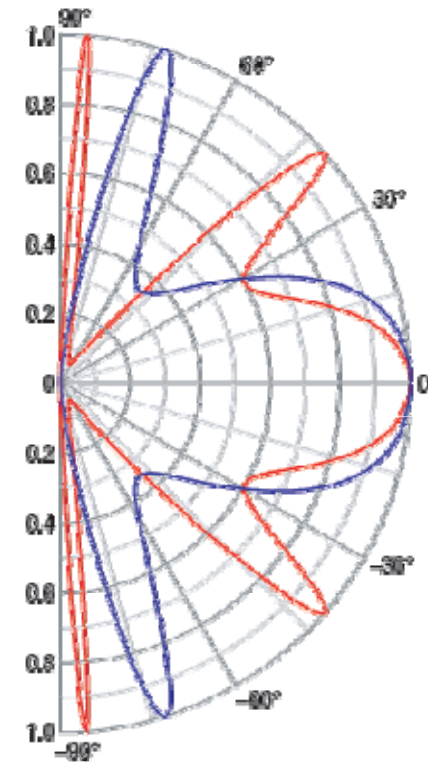
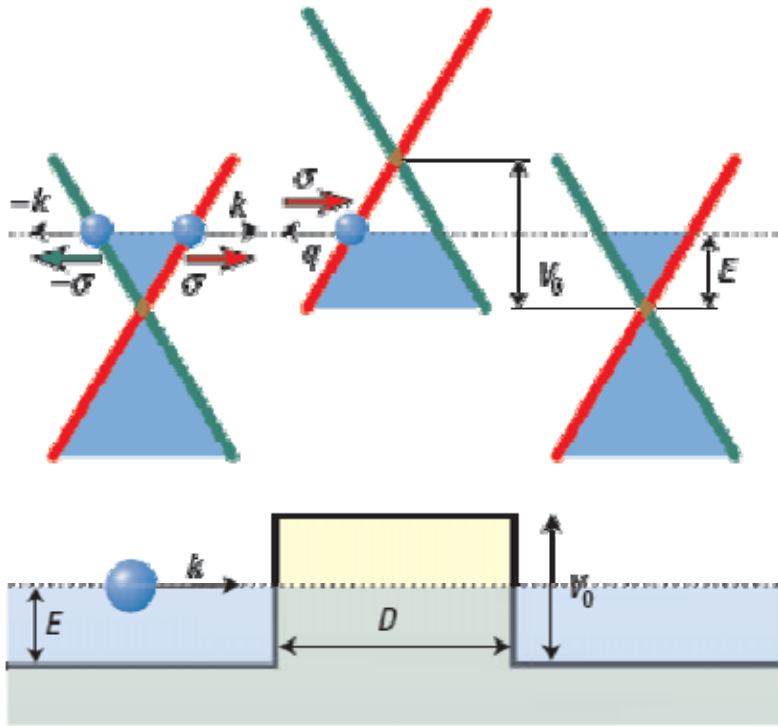


A. K. Geim & K. S. Novoselov
Nature Materials **6**, 183 (2007)

Klein tunneling:

Conservation of Pseudospin. (pseudospin flipping is prohibited)

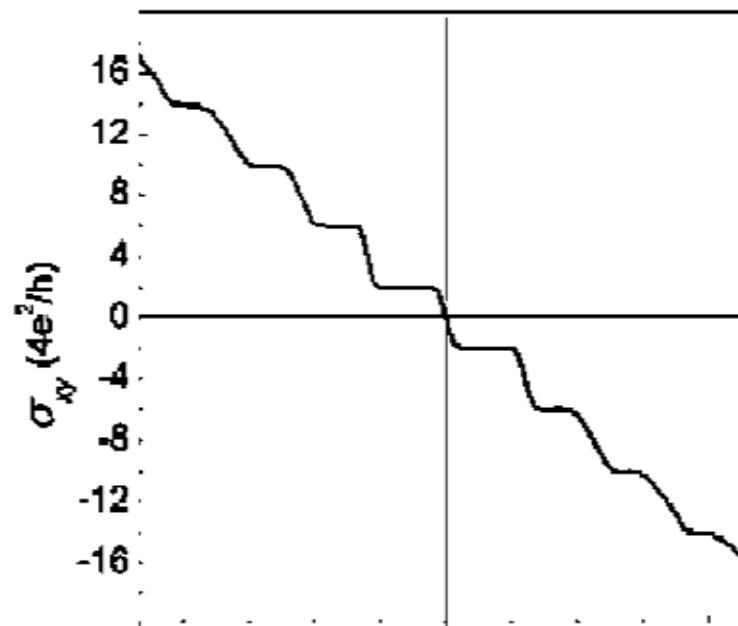
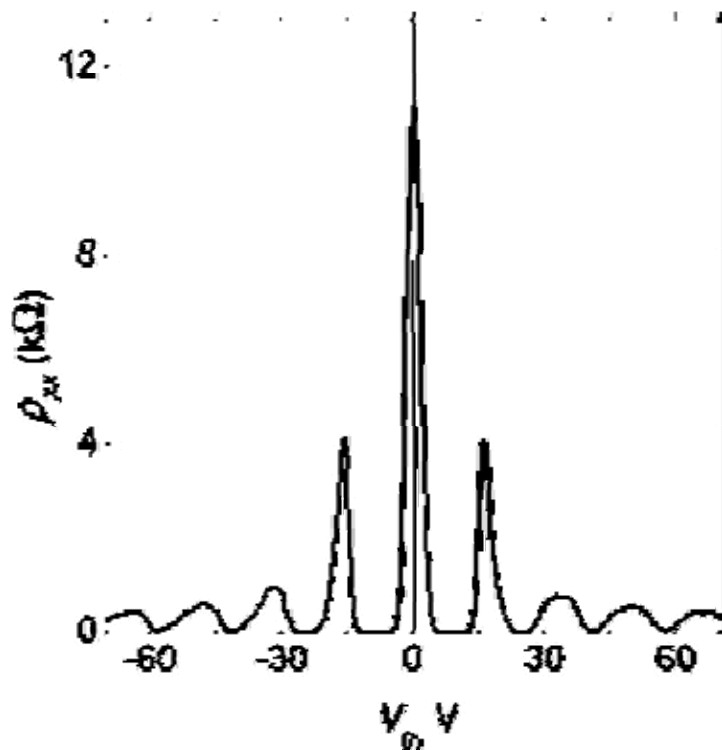
Weak electrostatic confinement.



M. Katsnelson et al *Nature Physics* 2, 620 (2006)

Quantum Hall effect:

T=4K; B=14T



K. Novoselov et al phys. Stat. sol. **244**, 4106 (2007)

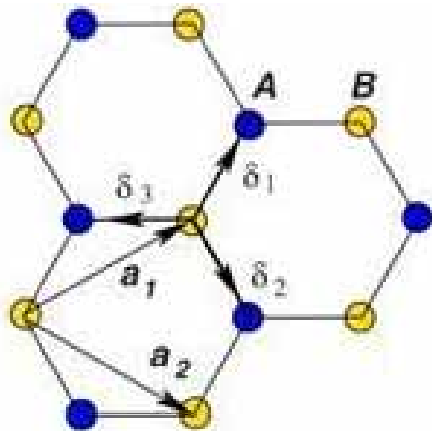
$$\sigma_{xy} = \pm 4e^2/h(N + 1/2)$$



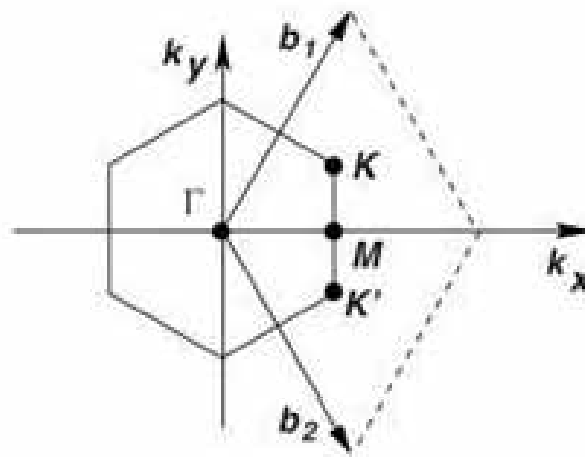
4=2(valley degeneracy)x2(spin degeneracy)

N: Landau Level

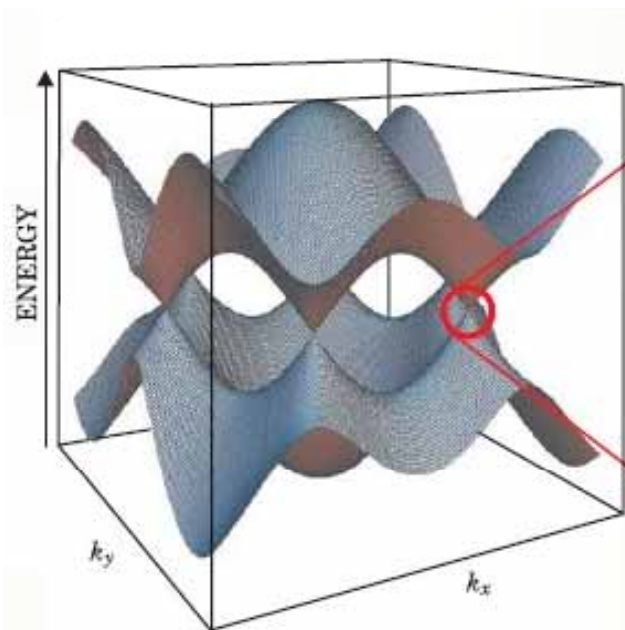
石墨烯的晶格結構



A/B sublattice



K/K' obey space inversion symmetry

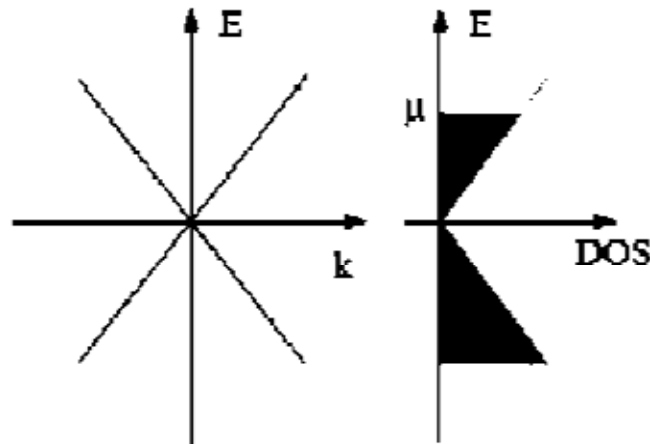


Linear dispersion nearby Dirac point

Zero bandgap semiconductor

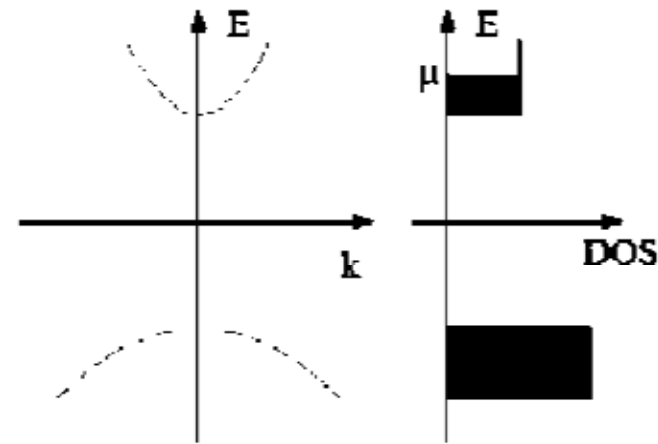
Electronic Properties

Graphene



- $\varepsilon(k) = \pm \hbar v_F k$, no gap
- $\text{DOS}(E) \sim E$

2DEG



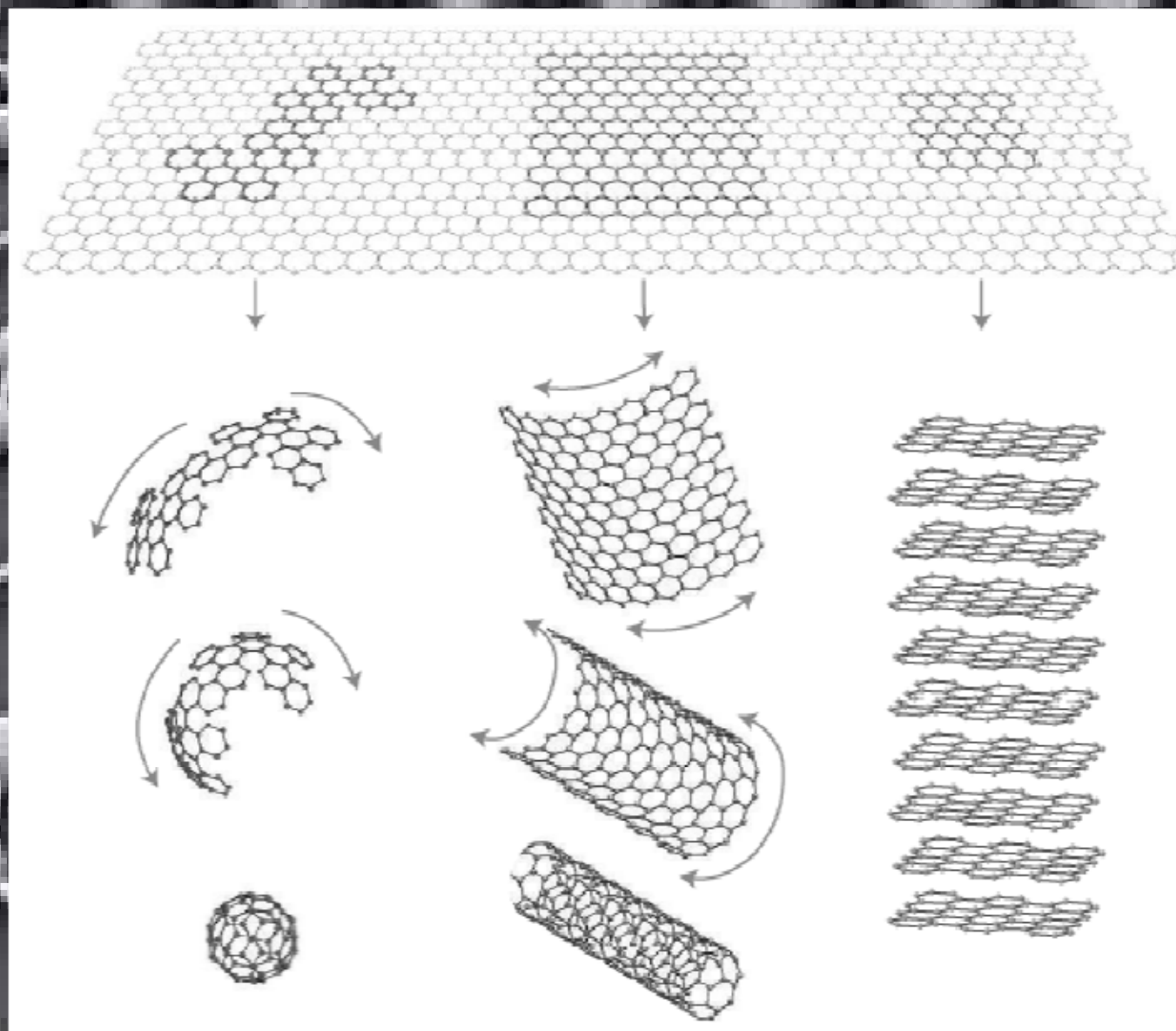
- $\varepsilon(k) = \hbar^2 k^2 / 2m^*$
- $\text{DOS}(E) = \text{const.}$

Dirac Fermion:

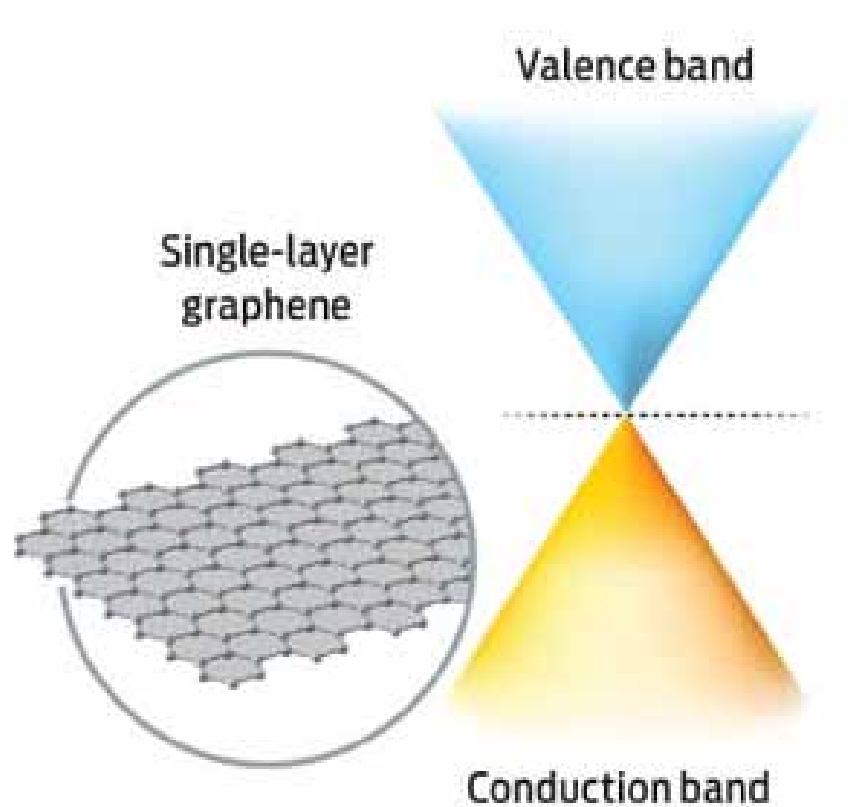
$$E_{Dirac}(k) = \pm \sqrt{\hbar^2 c^2 k^2 + m^2 c^4}$$

$$E_{Gra}(k) = \pm \hbar v_F k \quad m_{eff} = 0, v_F \sim 10^6 \text{ m/s}$$

石墨烯的衍生物

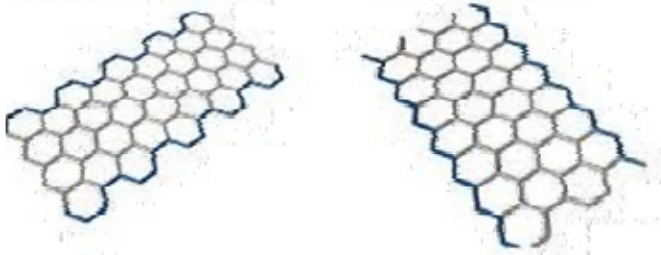


Replacement of Silicon in IC?

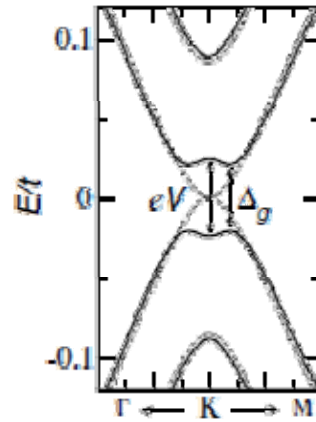


Need a gap (switch on/off) for transistor application.

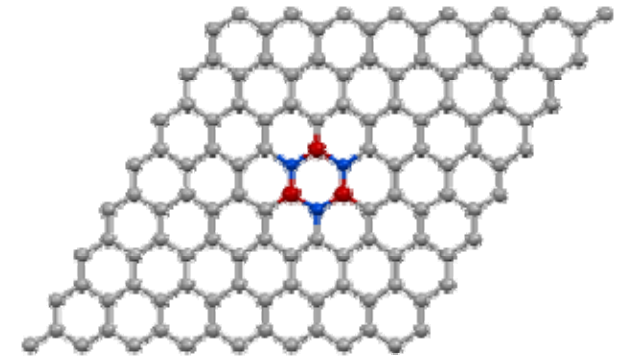
How to open a gap in graphene ?



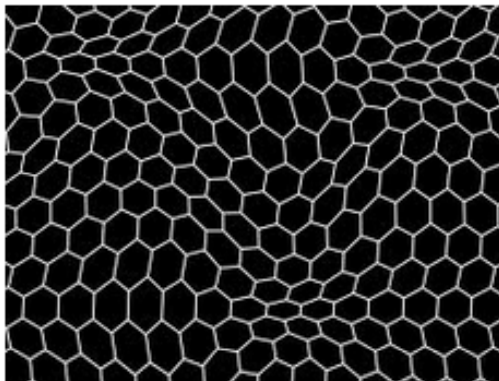
Nnanoribbon



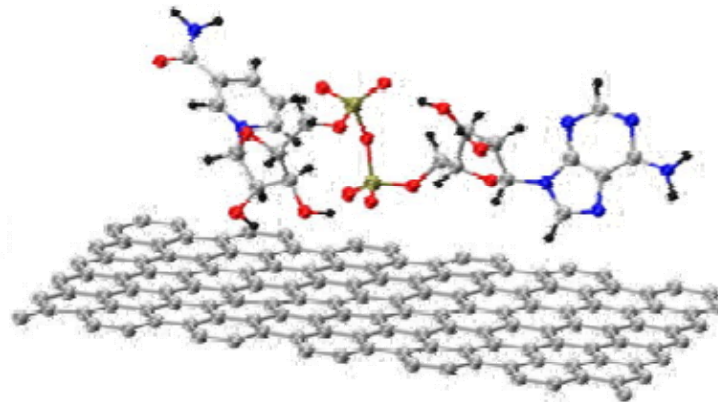
Apply bias on BLG



Chemical substitution

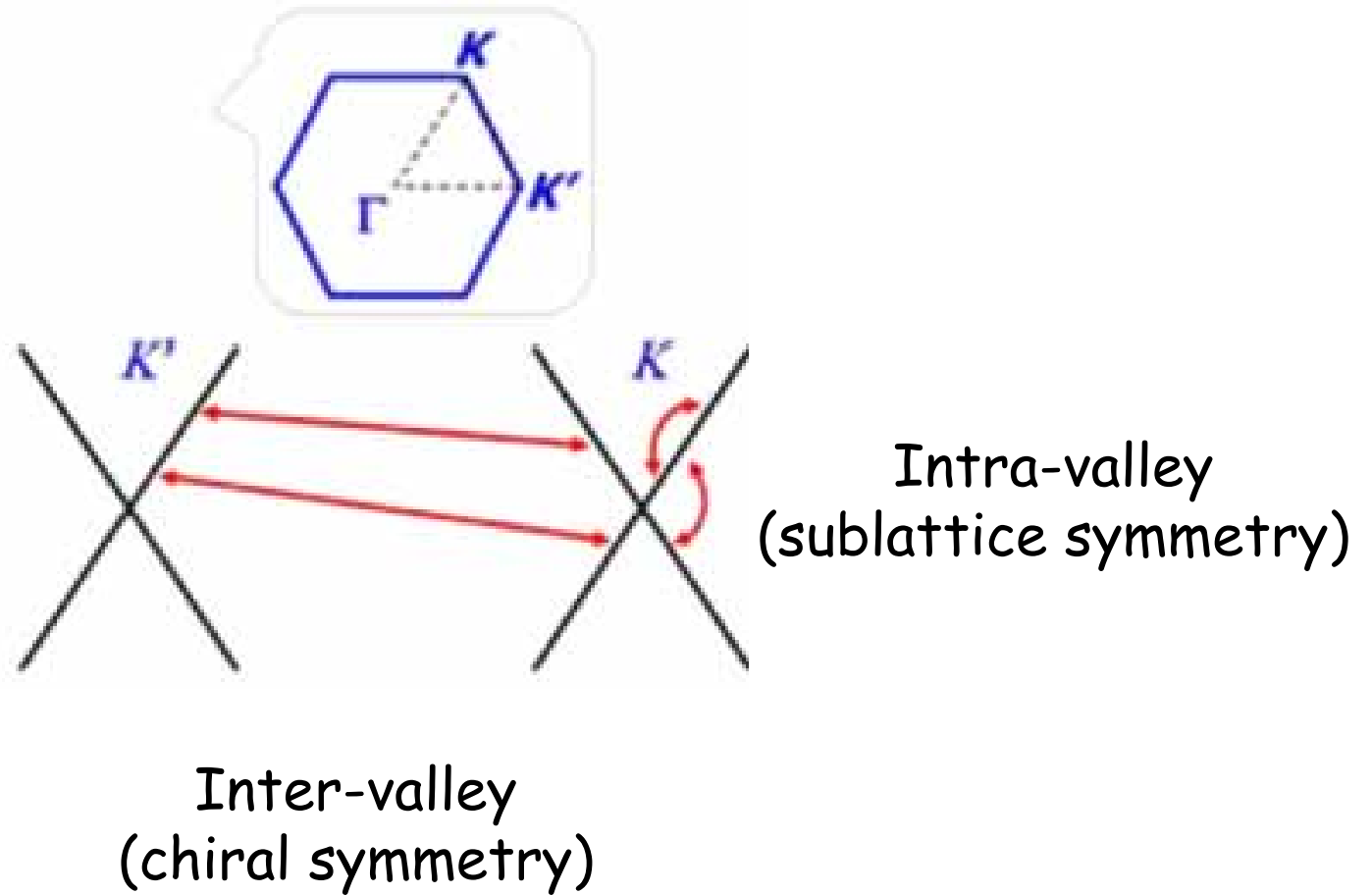


Strain

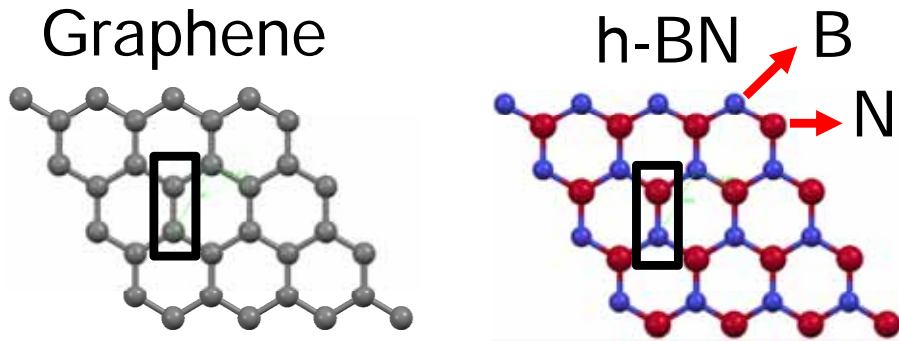


Molecule absorption

Gap opening mechanism



Gap opening due to Sublattice symmetry broken

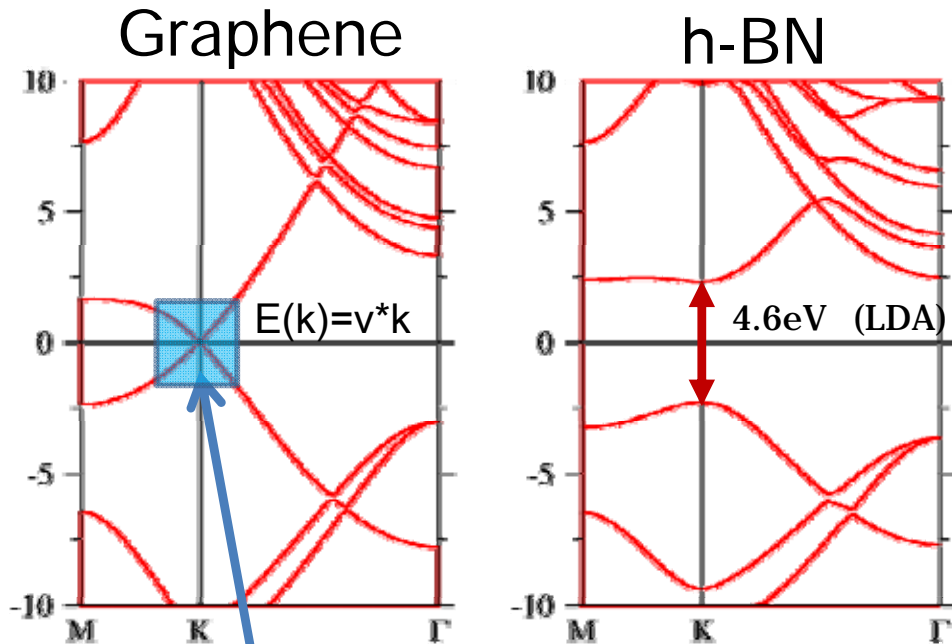


Sublattice Symmetry Broken

Lattice mismatch ~1.7%

$$E(\mathbf{k}) = \pm \sqrt{(\mathbf{v} \cdot \mathbf{k})^2 + (E_g/2)^2}$$

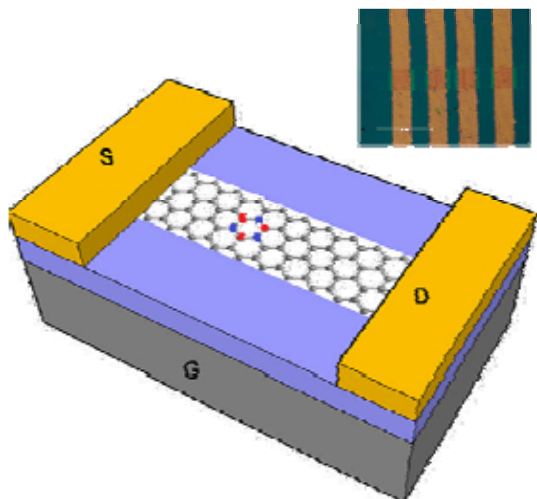
On-site energy difference at sublattices



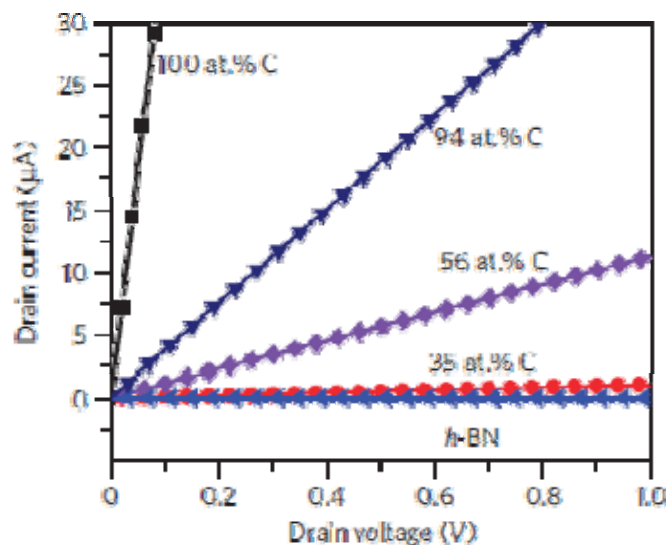
Gapless at Dirac point

Recent development of h-BNC applications

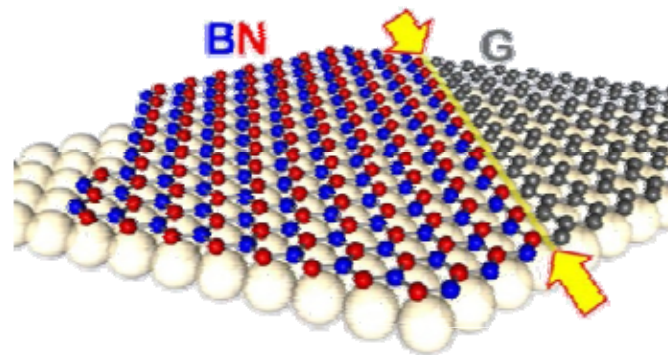
h-BNC FET (2010)



L. Ci et.al. Nature Mat. 2010

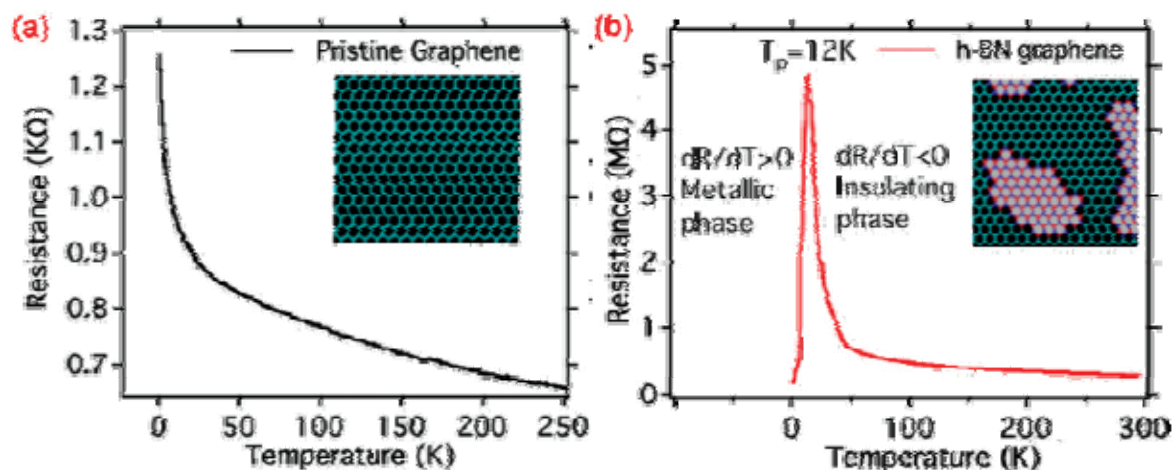


Lateral heterostructures



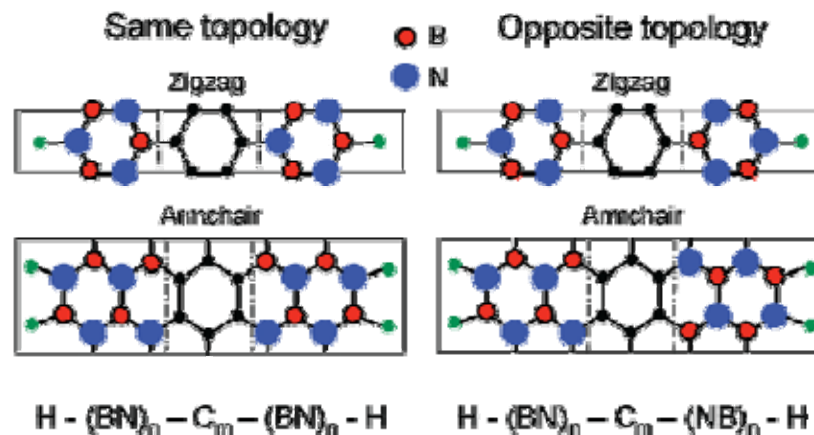
P. Sutter et. al. Nano. Lett. 2012
M. Levendorf et.al. Nature 2012

Anomalous insulator-metal transition (IMT)



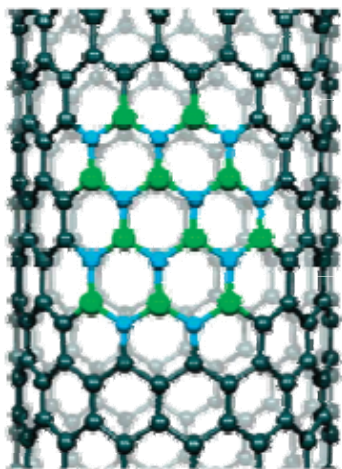
L. Song et.al. Phys. Rev. B 2012

1D transport



J. Jung et.al. Nano Lett. 2012

BN Domains embedded in Carbon Nanotubes



A strong tendency for the formation of BN pairs and, subsequently, BN hexagons within the carbon lattice. At higher doping concentrations, it is found that the formation of compact BN domains would be the most energetically favorable. The importance of a correct description of the BN/C frontier, due to the large border/surface ratio of small dopant domains.

Periodic Table of the Elements

Valence

<http://www.chemistry.about.com>

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About Chemistry

1A												3A					8A	
1 H Hydrogen											5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	2 He Helium		
2A												4A					7A	10
3 Li Lithium	4 Be Beryllium											13 Al Aluminum	14 Si Silicon	15 P Phosphorus	16 S Sulfur	17 Cl Chlorine	10 Ne Neon	
												5A					6A	18
11 Na Sodium	12 Mg Magnesium											31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	18 Ar Argon	
		3B	4B	5B	6B	7B	8B		1B	2B						36		
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton	
																	54	
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon	
																	86	
55 Cs Cesium	56 Ba Barium	57-71 Lanthanides	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon	
																	118	
87 Fr Francium	88 Ra Radium	89-103 Actinides	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Uut Ununtrium	114 Fl Flerovium					



Lanthanides

57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
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Actinides

89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium
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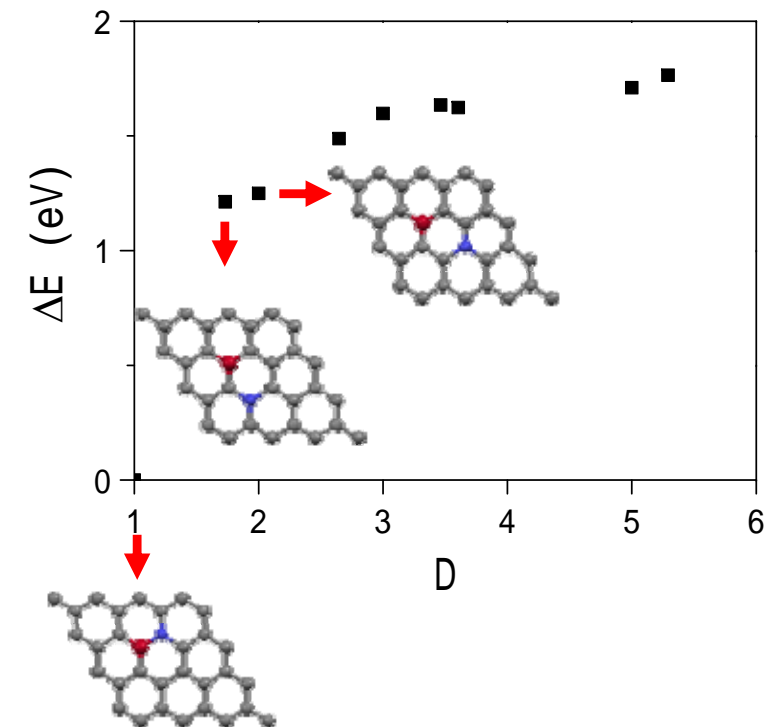
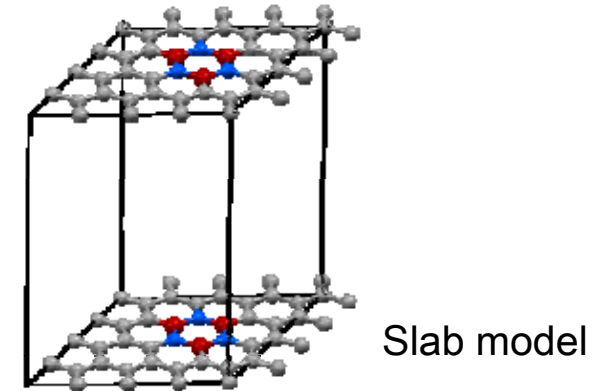
Gap opening due to Sublattice symmetry broken

B B (one electron less than C): p-dopant

N N (one electron less than C): n-dopant

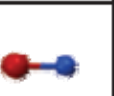
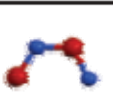
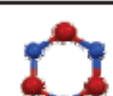
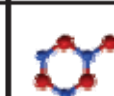
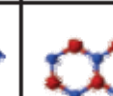
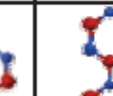

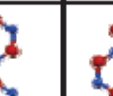
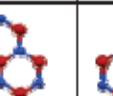
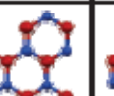
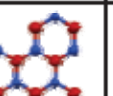
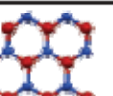
B — **N** B-N (Effective dipole)

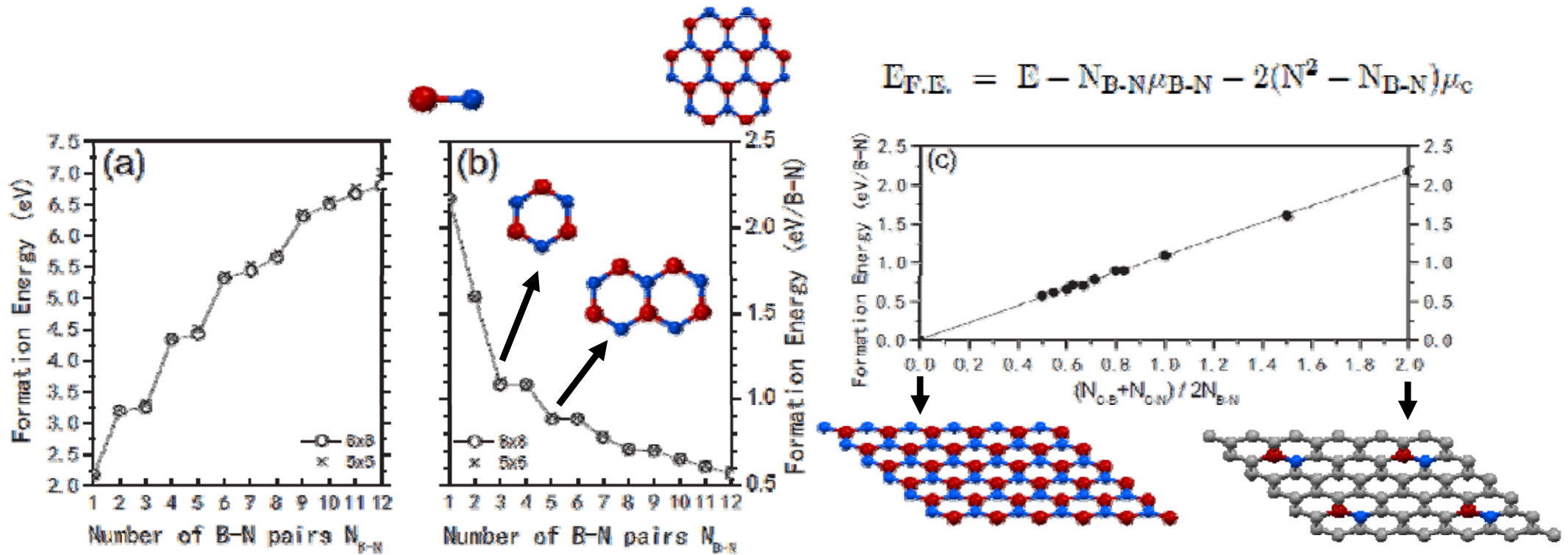
1st Principles calculation (PAW-LDA)



Energetics structures of B-N clusters in h-BNC : Phase Separation

Optimized structures of B-N dopants on 8 × 8 supercell

N_{B-N}	1	2	3	4	5	6	7	8	9	10	11	12
$\frac{N_{C-B} + N_{C-N}}{2N_{B-N}}$	2.00	1.50	1.00	1.00	0.80	0.83	0.71	0.63	0.67	0.60	0.55	0.50
Structure												

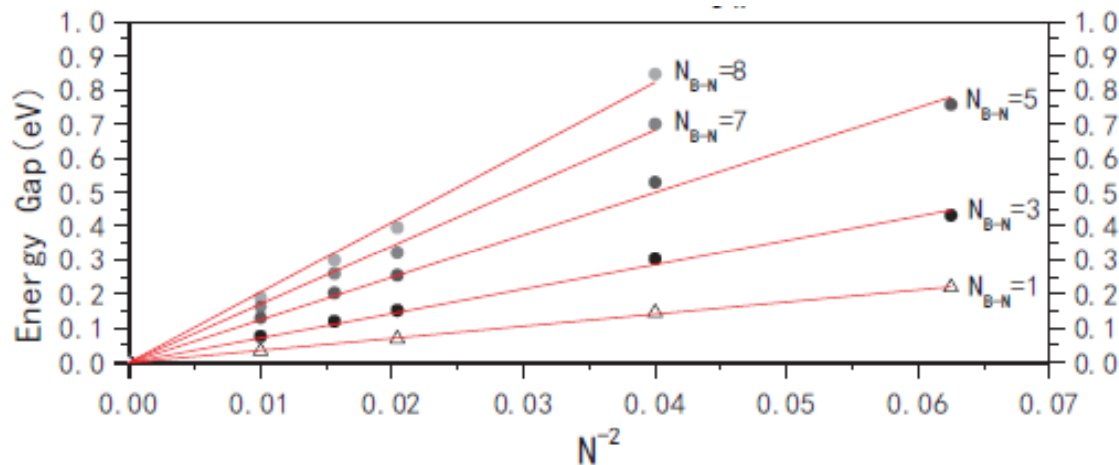
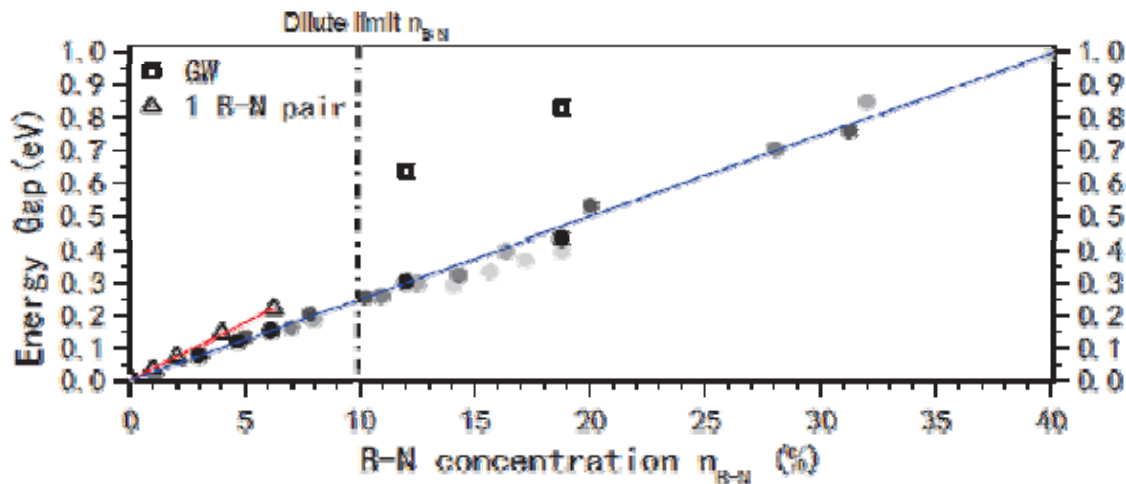


Formation Energy as number of B-N pair of h-BNC

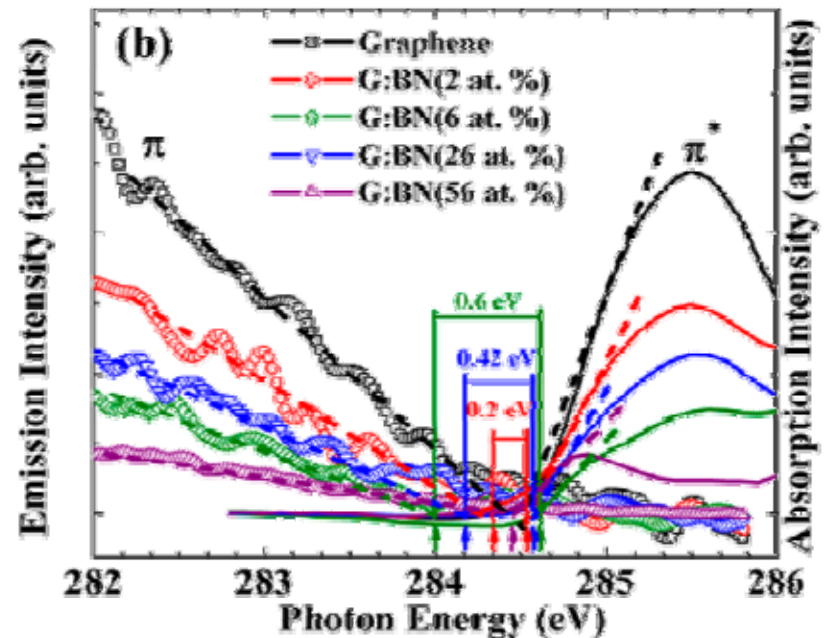
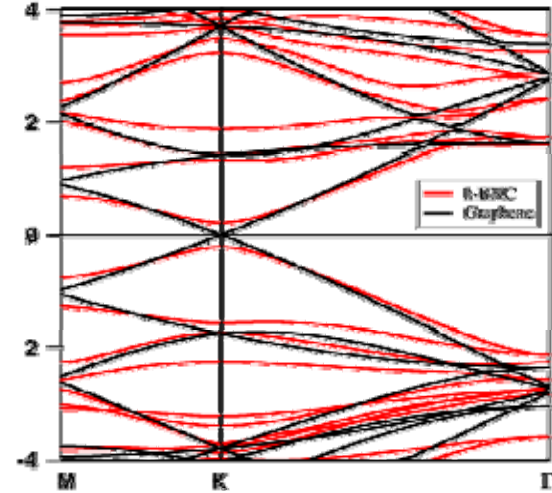
Phase separation is energetic.
C.-P. Chuu et.al. 2012

Gap opening of B-N clusters in h-BNC : **Low** concentration

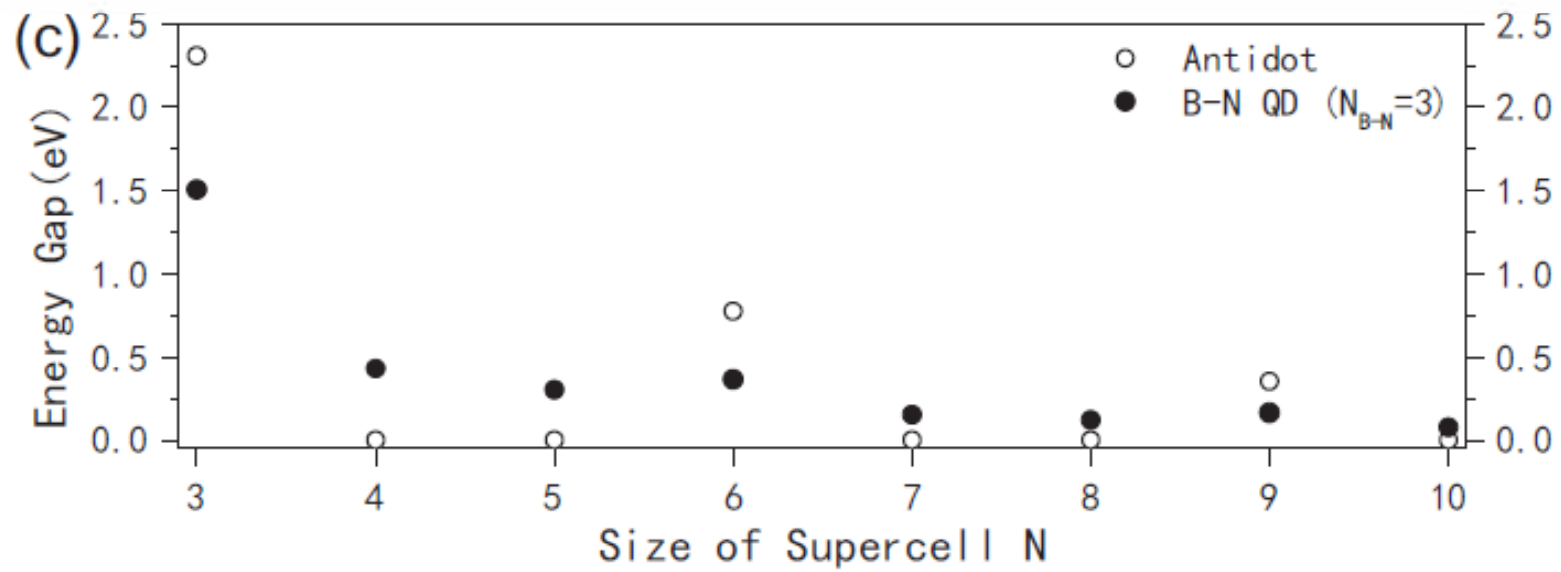
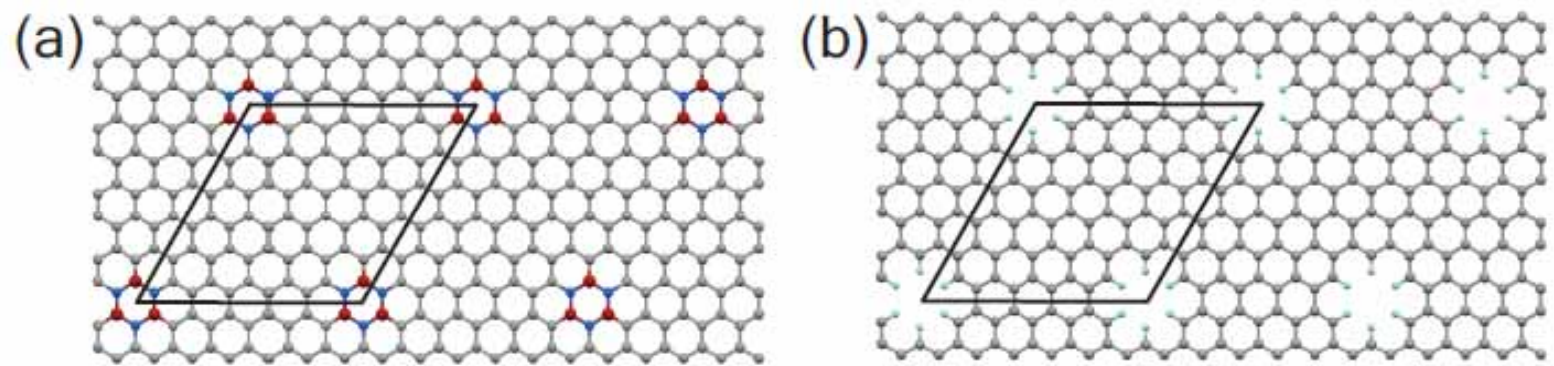
Band Structure of 4x4 h-BNC and pristine graphene



Energy gap vs B-N concentration of h-BNC
C.-P. Chuu et.al. 2012

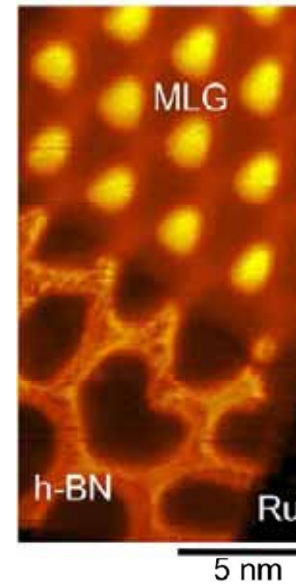
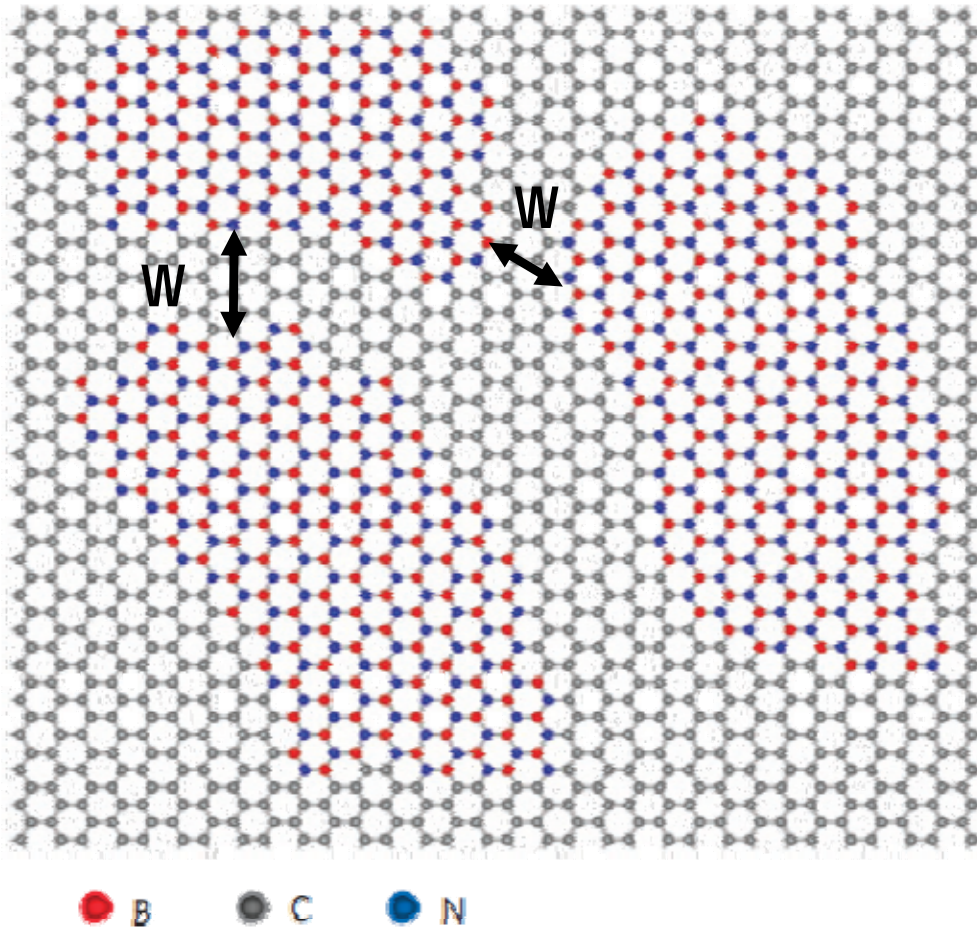


XANES (X-ray Absorption Near-Edge Structure)
C-K Chang et.al. ACS Nano 2013

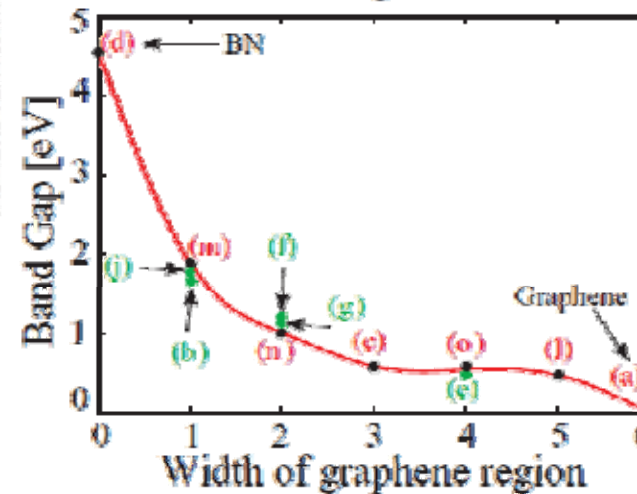


Gap opening of B-N clusters in h-BNC : High concentration

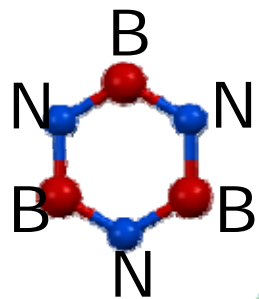
Phase Separation



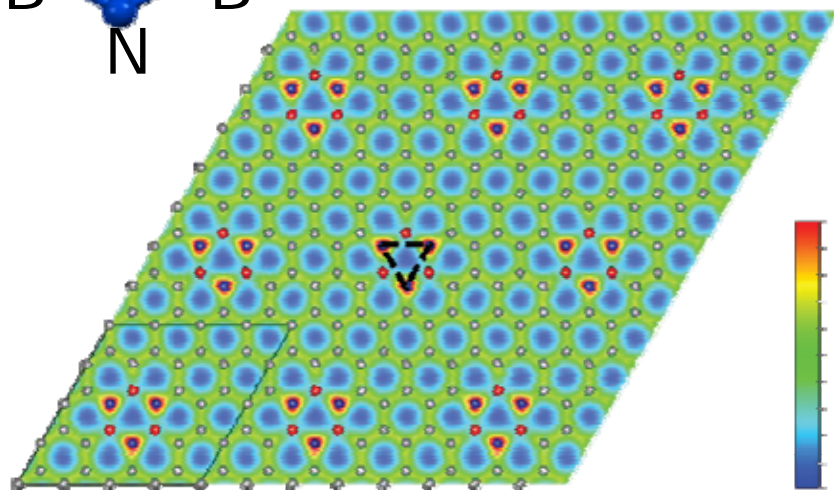
STM
P. Sutter et. al. Nano. Lett. 2011



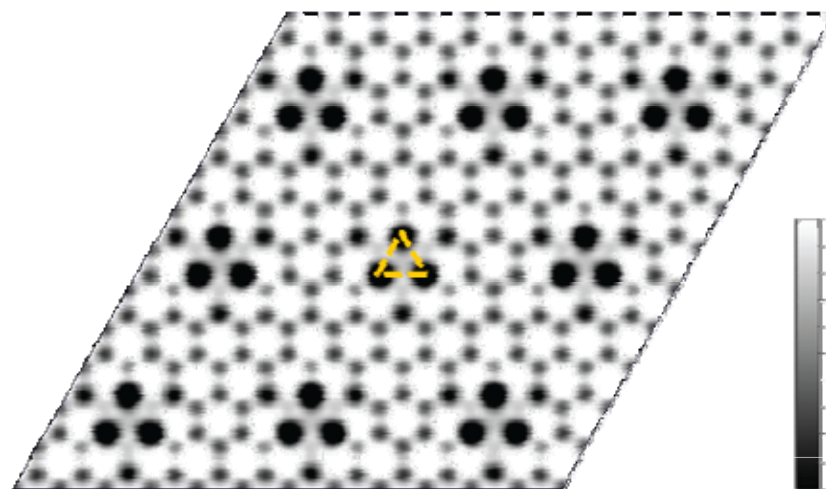
P. Shinde and V. Kumar, Phys. Rev. B 2011



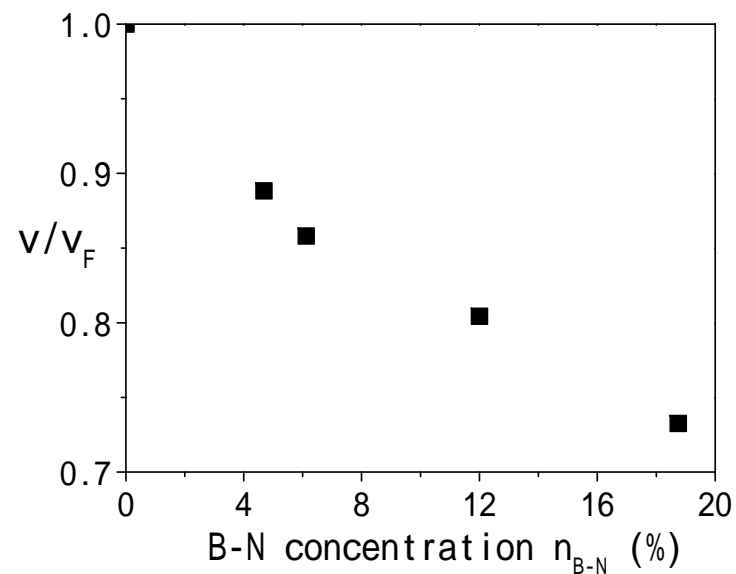
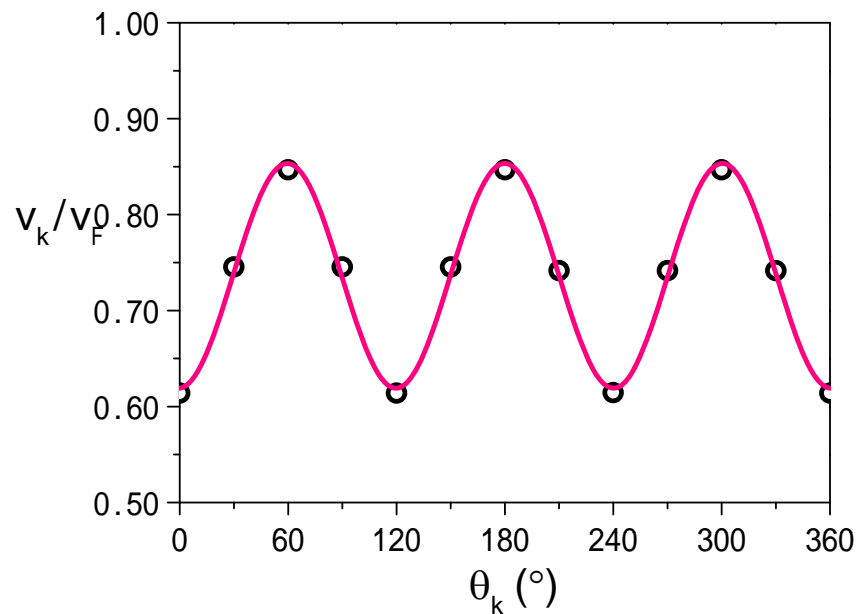
Anisotropy behavior of Dirac fermions in h-BNC



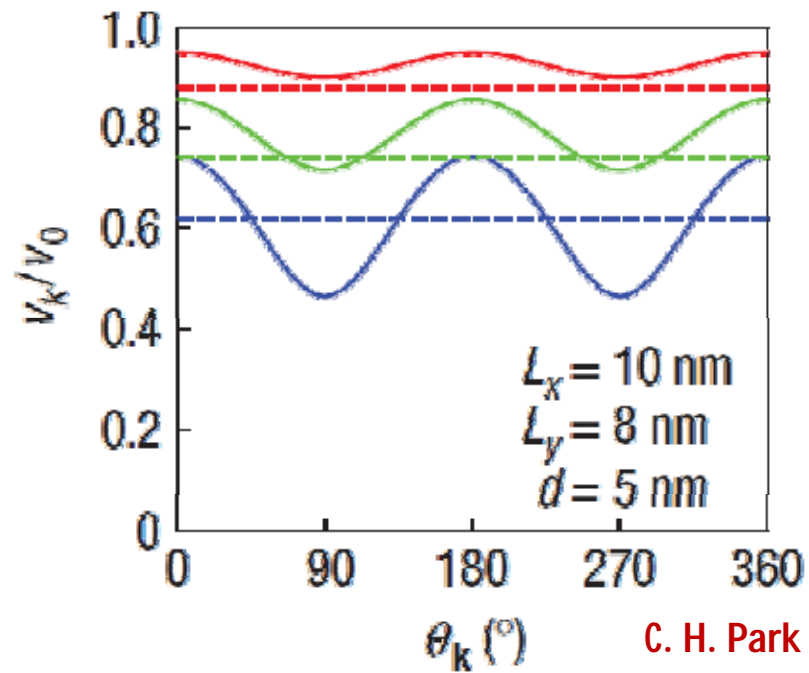
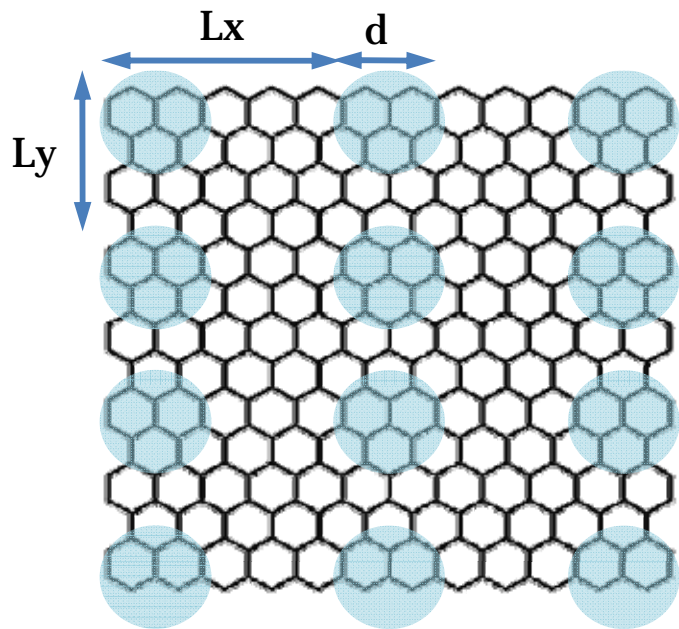
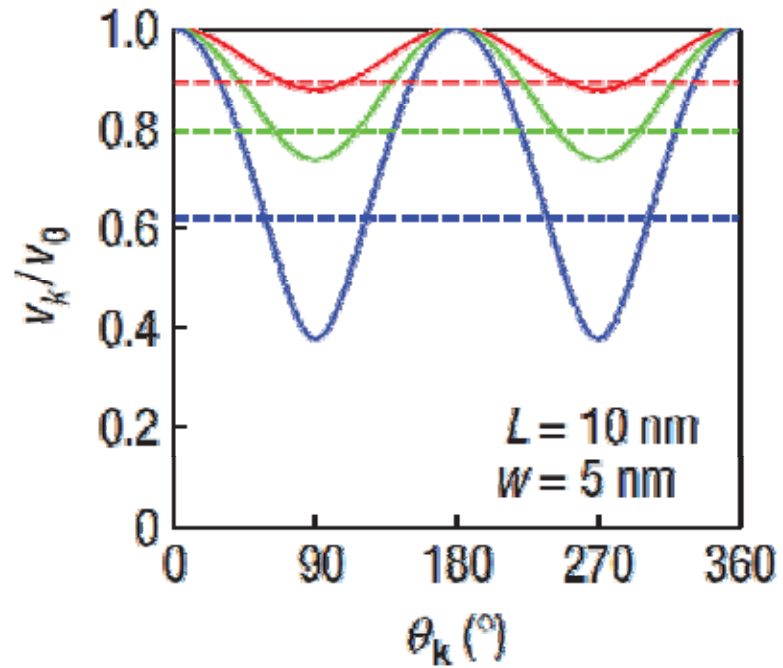
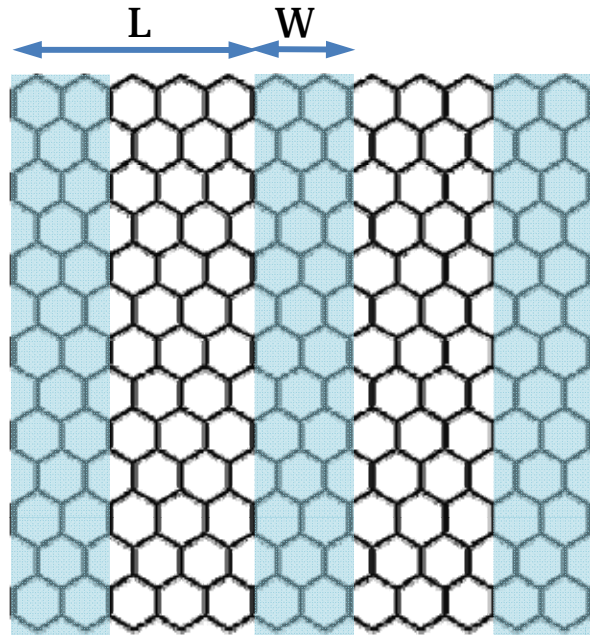
Charge density



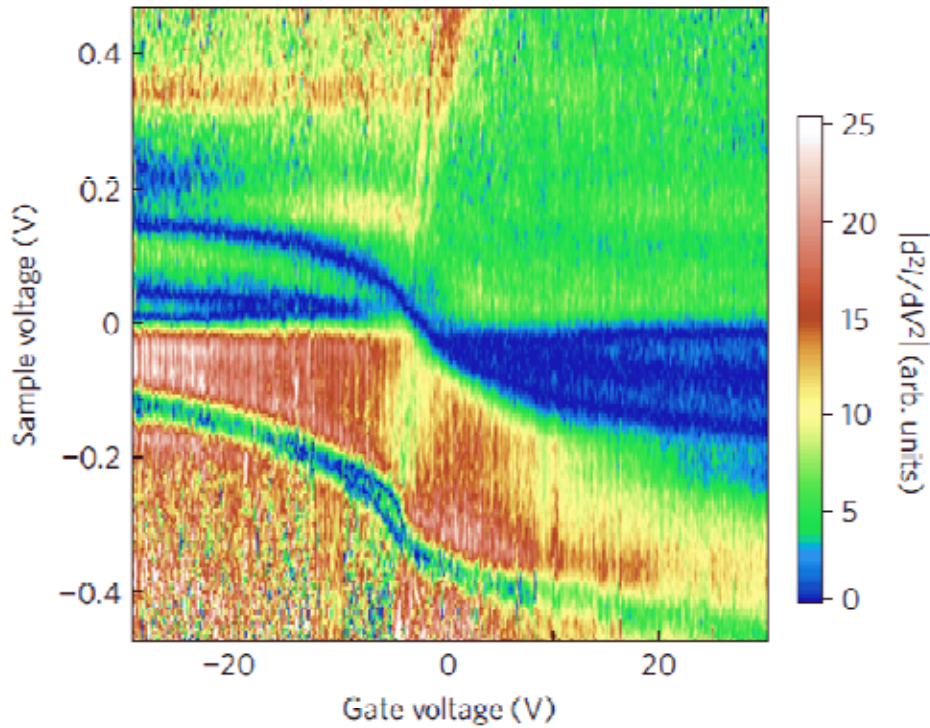
Local potential surface



Anisotropy behavior of Dirac fermions in graphene under periodic potential



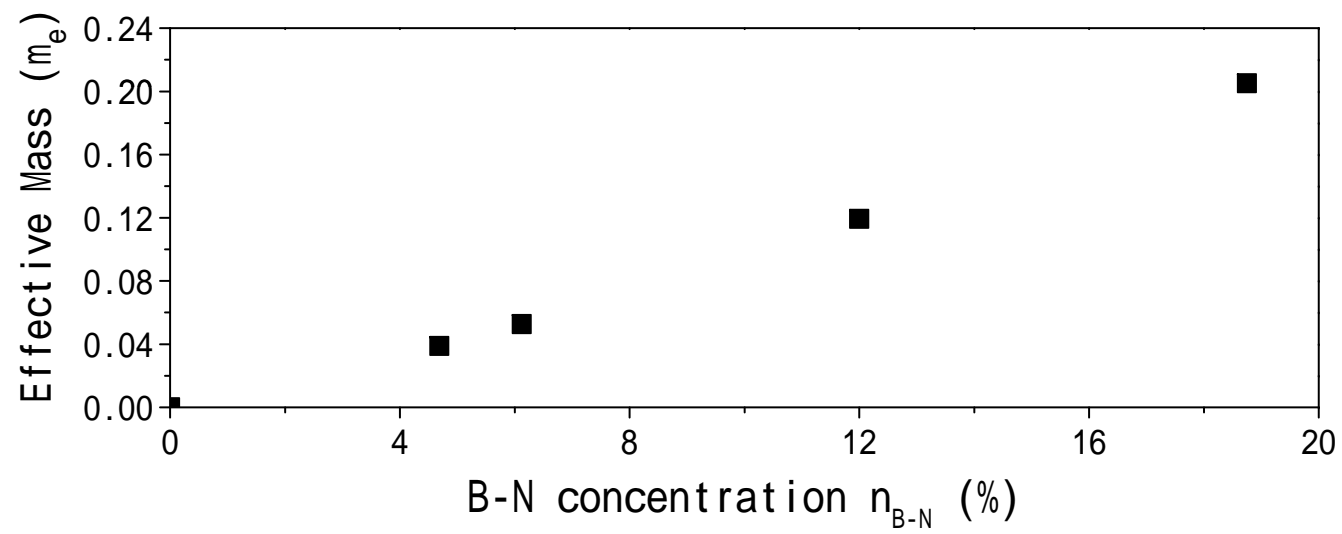
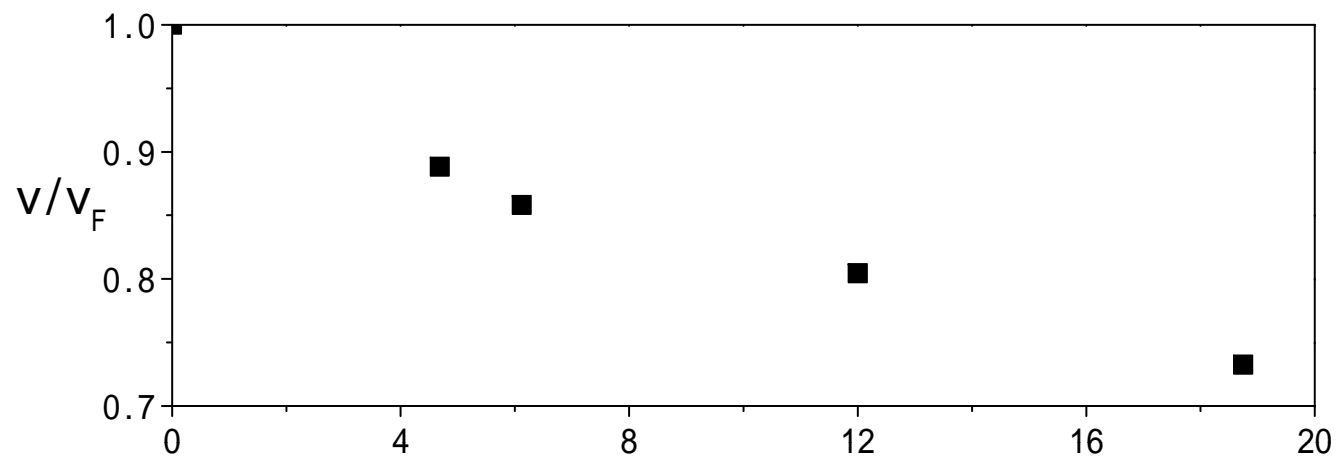
Emergence of superlattice Dirac points in graphene on h-BN



$$E_D = \hbar v_F \sqrt{2\alpha\pi(V_g - V_0)/g_v}$$

g_v : Valley degeneracy

V_g / V_0 : Gate/offset voltage



Summary

1. The phase separation between B-N domain and graphene is energetic by minimizing the number of C-B and C-N bonds at boundary .
2. The gap induced by symmetry broken is studied, a linear relation as B-N concentration at dilute limit is verified from experiment by XANES. At high doping level, the quantum confinement effect is dominant.
3. The anisotropy behavior of Dirac mass fermion is observed in h-BNC system, analogy to graphene under periodic potential , caused by periodic array of B-N dopants.

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